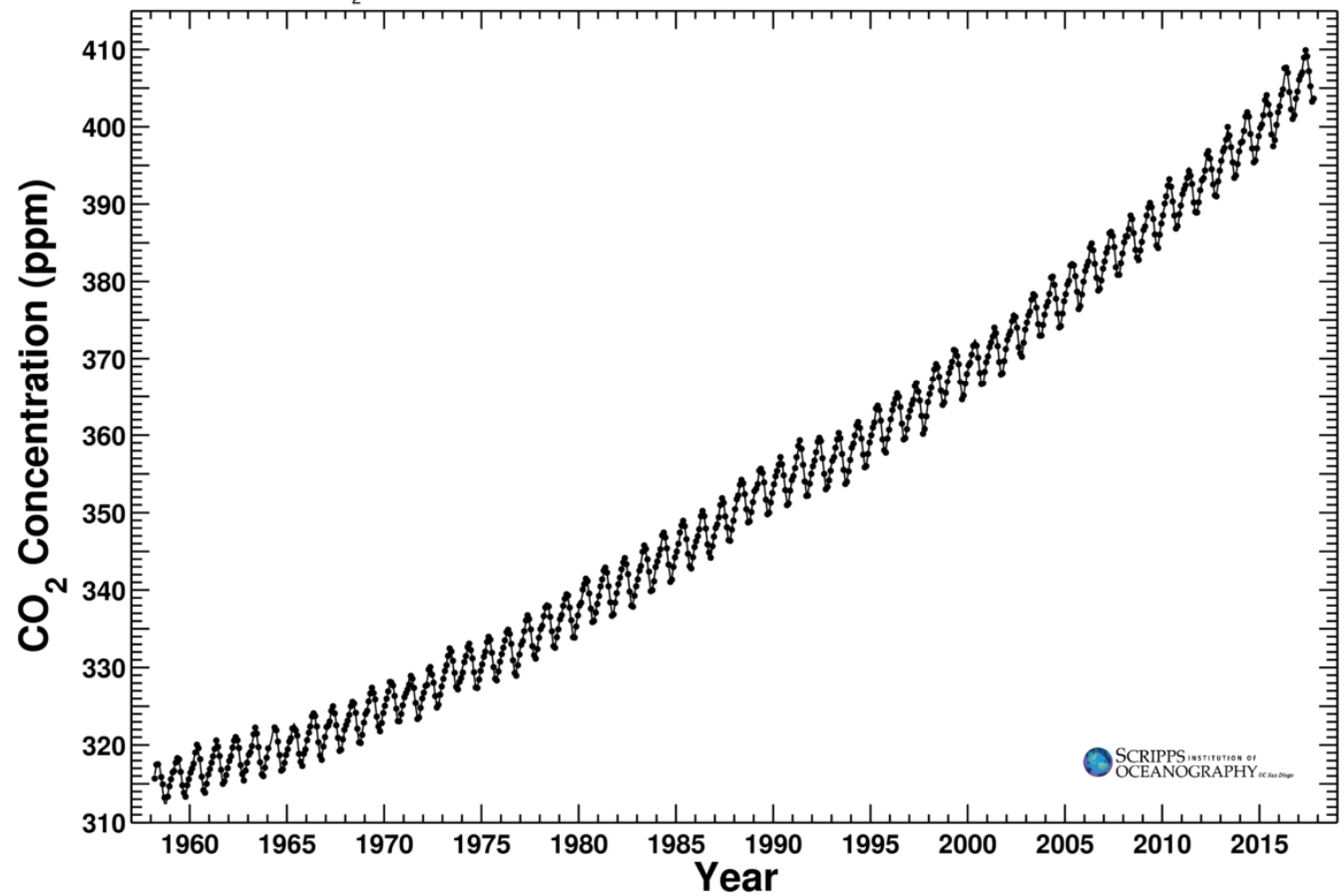


# The global carbon cycle

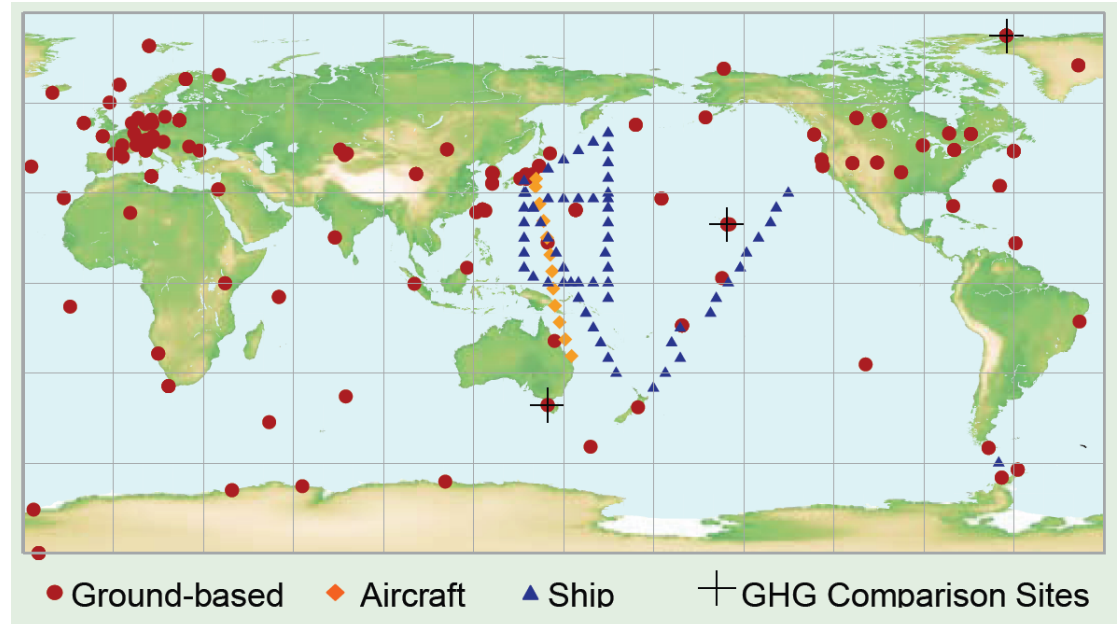
Heather Graven

## Mauna Loa Observatory, Hawaii Monthly Average Carbon Dioxide Concentration

Data from Scripps CO<sub>2</sub> Program Last updated November 2017



# Atmospheric measurements are conducted all over the world

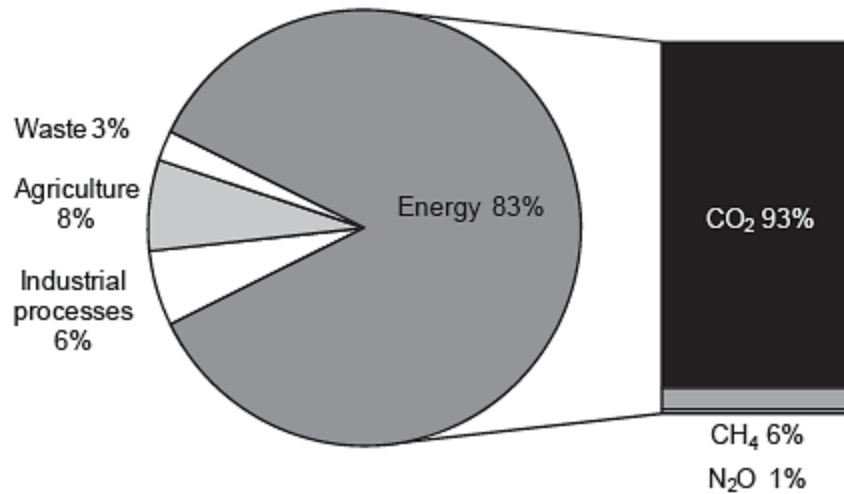


Images courtesy SIO, NSF, NASA, NERC

# GHG emissions have been growing rapidly

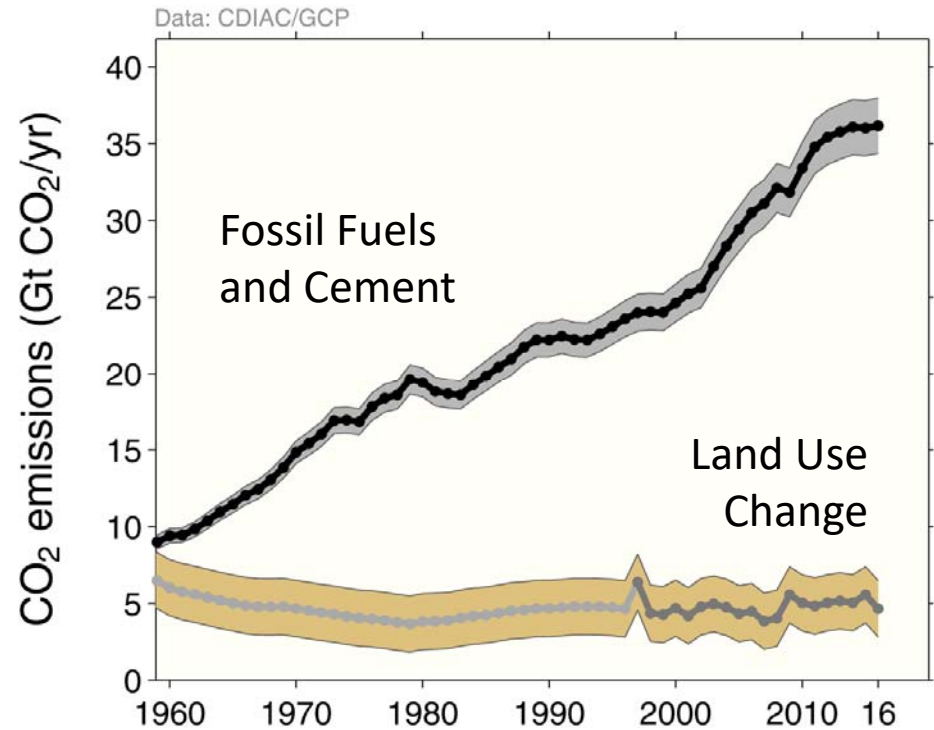
CO<sub>2</sub> emissions from fossil fuel combustion for energy are the largest source of GHGs by far

Anthropogenic GHG Emissions



\* Based on Annex I data for 2011; without Land Use, Land-Use Change and Forestry, and with Solvent Use included in Industrial Processes and "other" included with waste.

Source: UNFCCC.



Global Carbon Project

# Units

**1 Gigatonne (Gt) = 1 billion tonnes =  $1 \times 10^{15}$  g**

**1 ppm CO<sub>2</sub> = 7.8 Gt CO<sub>2</sub> = 2.1 Gt C**

## Examples

Emissions from burning 1 litre of petrol: 2.3 kg CO<sub>2</sub>

Mass of a car: 1000 kg = 1 tonne

Global per capita fossil fuel emissions in 2013: 5 tonnes CO<sub>2</sub> per person

UK per capita household waste in 2012-13: 420 kg per person

# Our energy system is heavily reliant on fossil fuels

Renewables like solar and wind energy have been growing rapidly but they are still a tiny fraction of total energy supply

Figure 1: 2014 fuel shares in world total primary energy supply

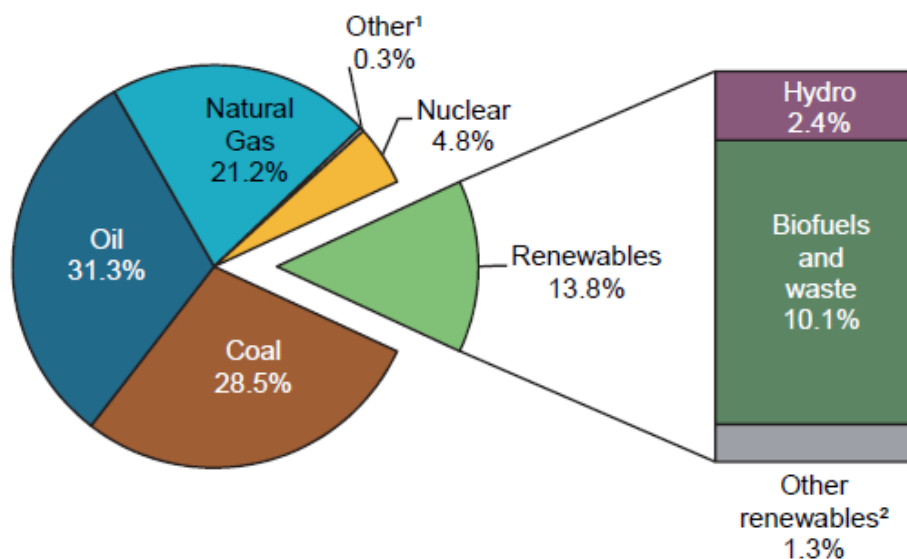
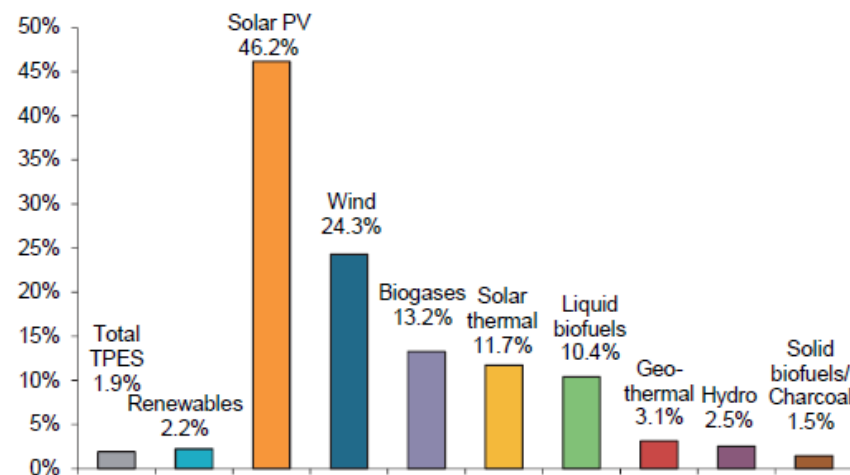


Figure 3: Annual growth rates of world renewables supply from 1990 to 2014



IEA Renewables Information (2016 edition)

Estimates of fossil fuel CO<sub>2</sub> emissions are based on accounting of activities according to e.g. IPCC guidelines



$$\text{Emissions} = \text{Activity Data} \times \text{Emission Factor}$$

$$\text{CI} = \frac{\text{Carbon Content (g C / g fuel)} \times \text{Combustion Efficiency (\%)} \times \frac{44 \text{ g CO}_2}{12 \text{ g C}}}{\text{Energy Density (kWh / g fuel)}}$$

Full life-cycle Carbon Intensity also depends on mining, transport, transmission, other GHGs, etc

### Indicative Carbon Intensity by Fuel

Fuel	gCO <sub>2</sub> kWh <sup>-1</sup>
Coal*	800 - 1000
Peat	740
Petrol / Diesel / Fuel Oil	700
Kerosene	640
Natural Gas*	400

\* Main fossil fuels used for electricity

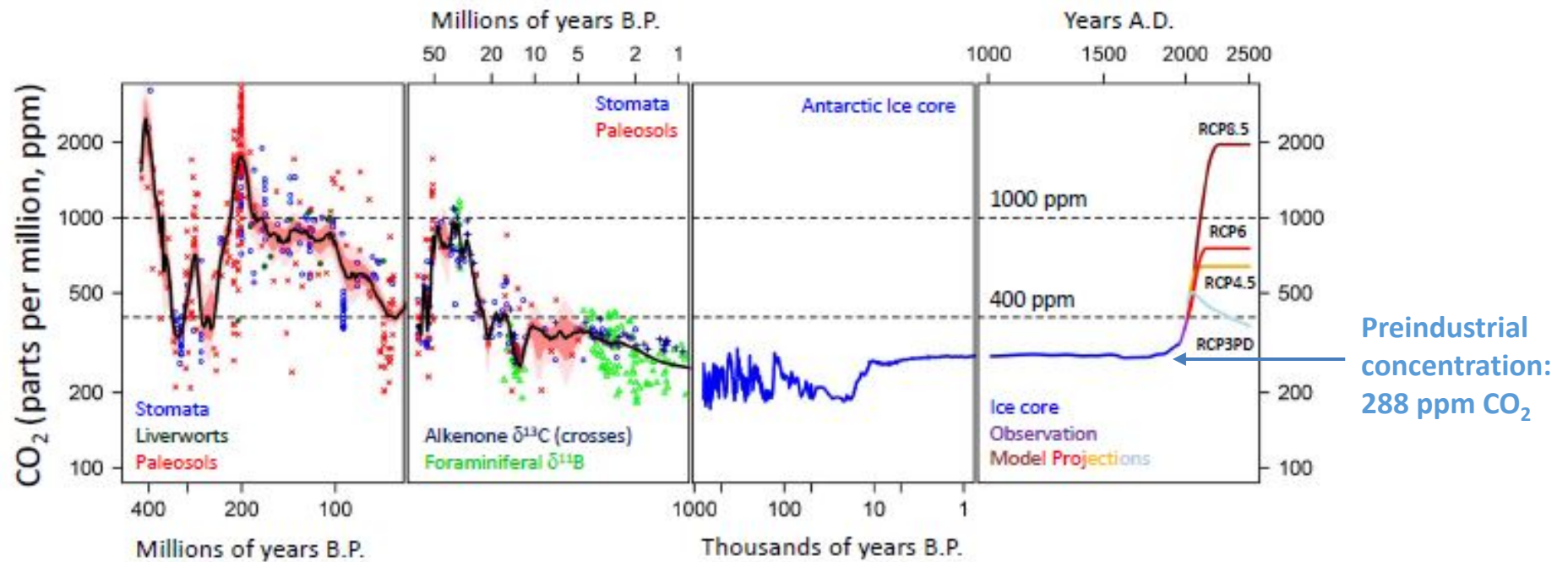
### Carbon Intensity of electricity by country, 2009-11

Country	gCO <sub>2</sub> kWh <sup>-1</sup>
China	780
US	500
UK	450
France	70
Norway	10

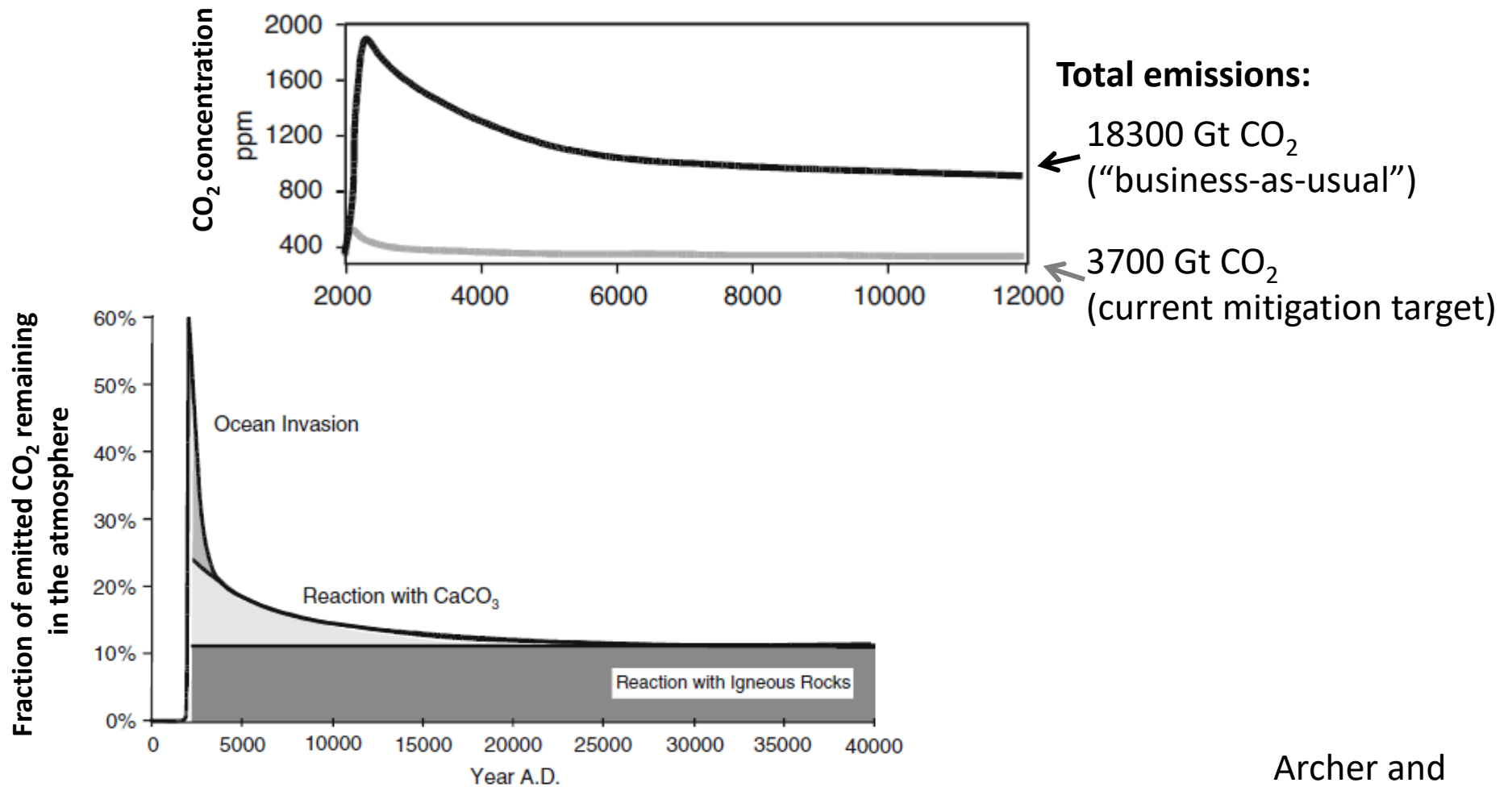
IEA, 2013



# Atmospheric CO<sub>2</sub> concentration has varied over Earth's history



# Atmospheric CO<sub>2</sub> remains elevated for millennia because some removal processes are slow

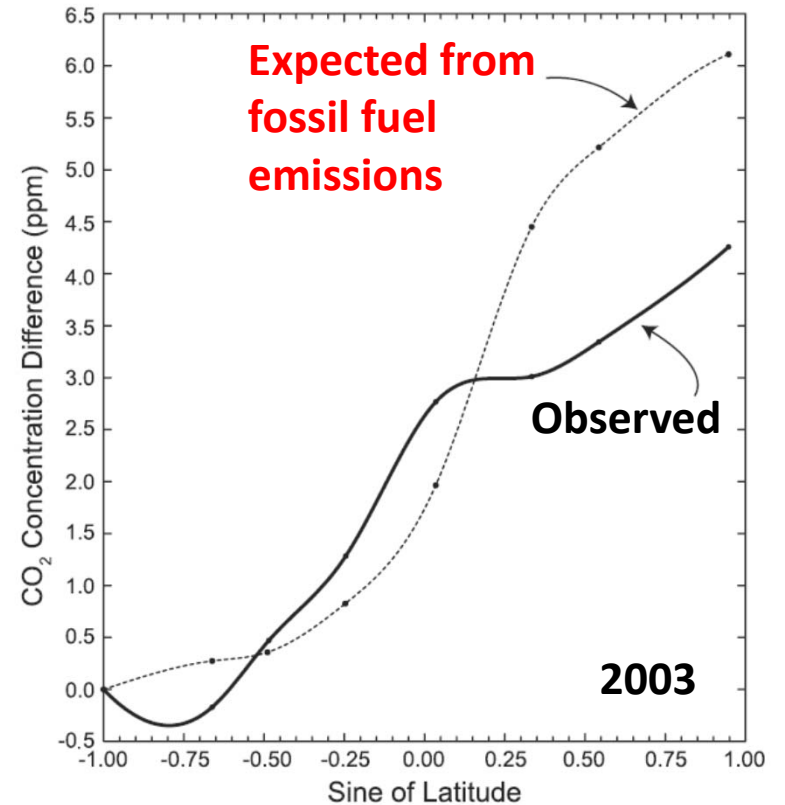
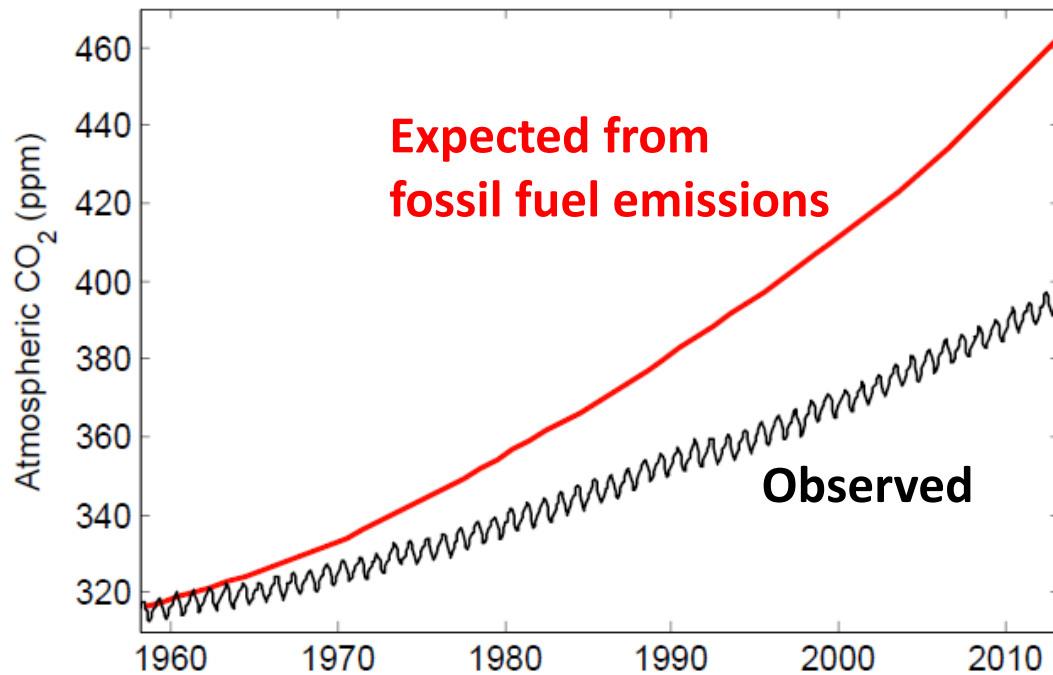


Archer and Brovkin 2008

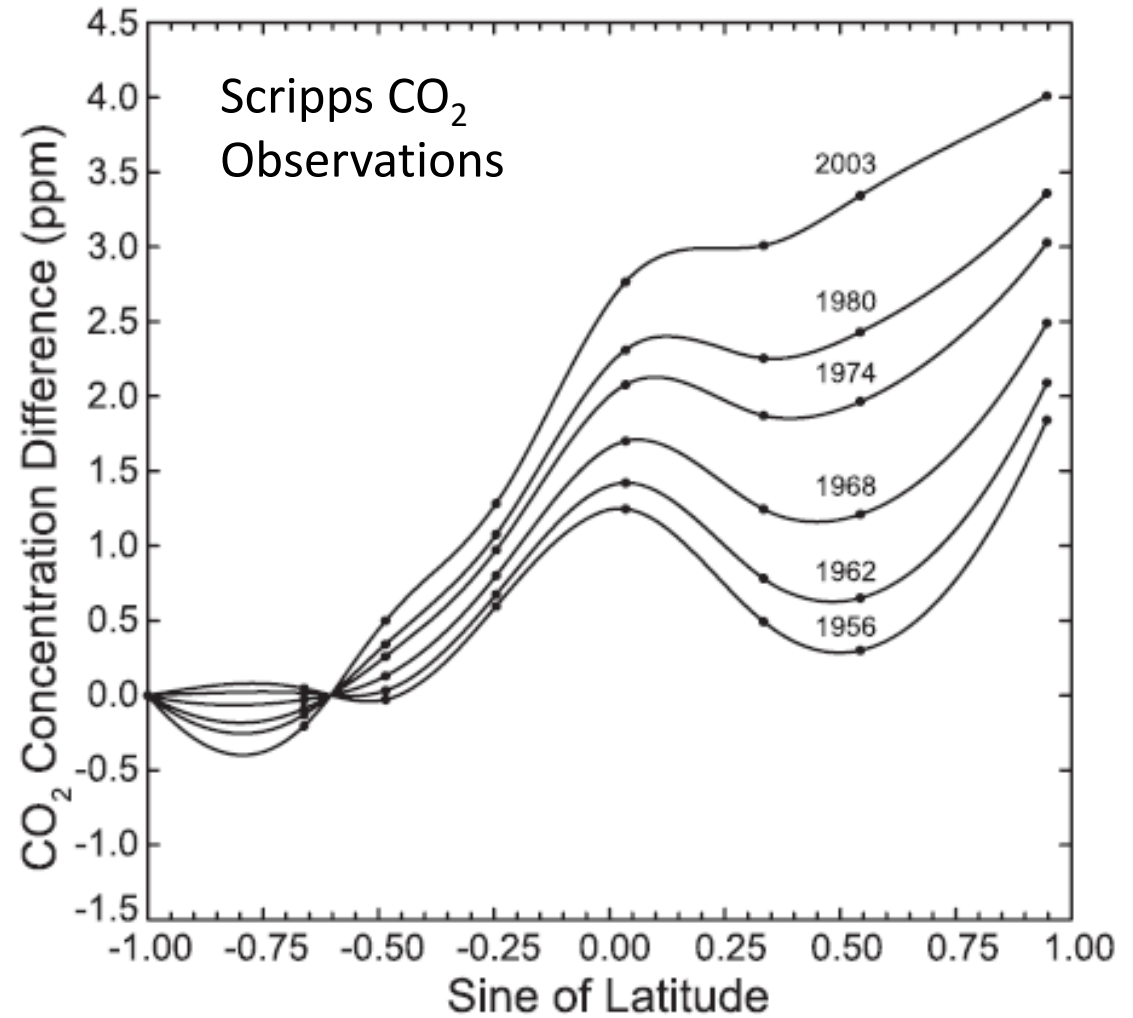
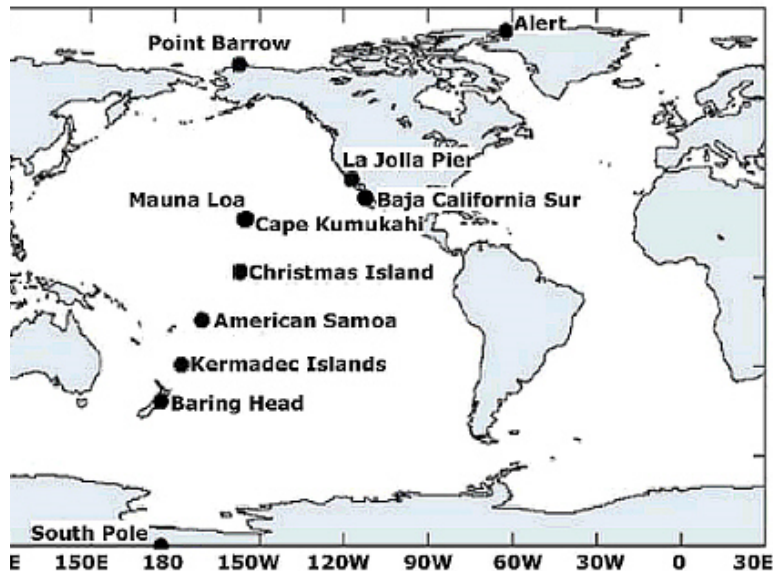


# Atmospheric CO<sub>2</sub> integrates sources and sinks

Interhemispheric gradient of atmospheric CO<sub>2</sub> is less than expected from emissions due to net CO<sub>2</sub> uptake by the oceans and by land plants in the Northern Hemisphere



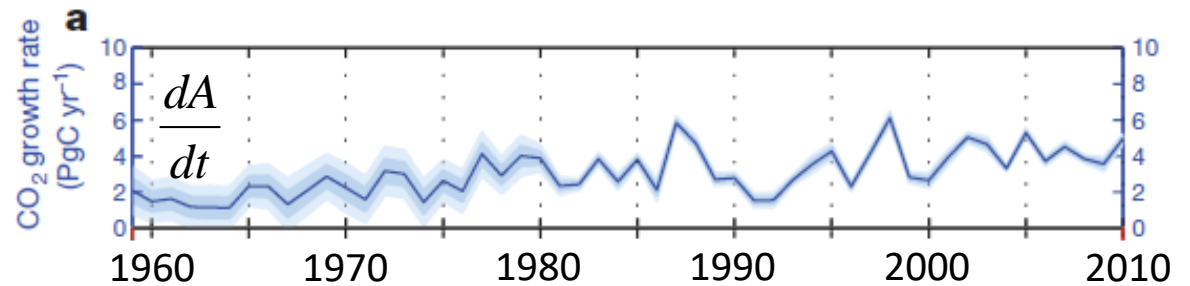
# The spatial distribution of CO<sub>2</sub> is evolving due to CO<sub>2</sub> sources and sinks



Keeling et al. 2011

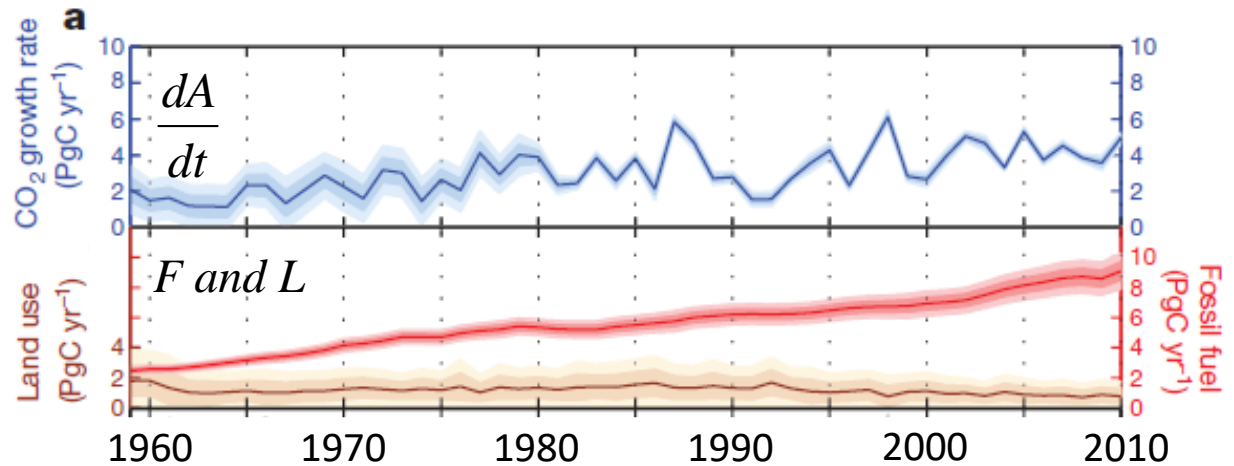
# Atmospheric CO<sub>2</sub> integrates net sources and sinks

$$\frac{dA}{dt} = F + L - O - B$$



# Atmospheric CO<sub>2</sub> integrates net sources and sinks

$$\frac{dA}{dt} = F + L - O - B$$



Airborne fraction: 45%

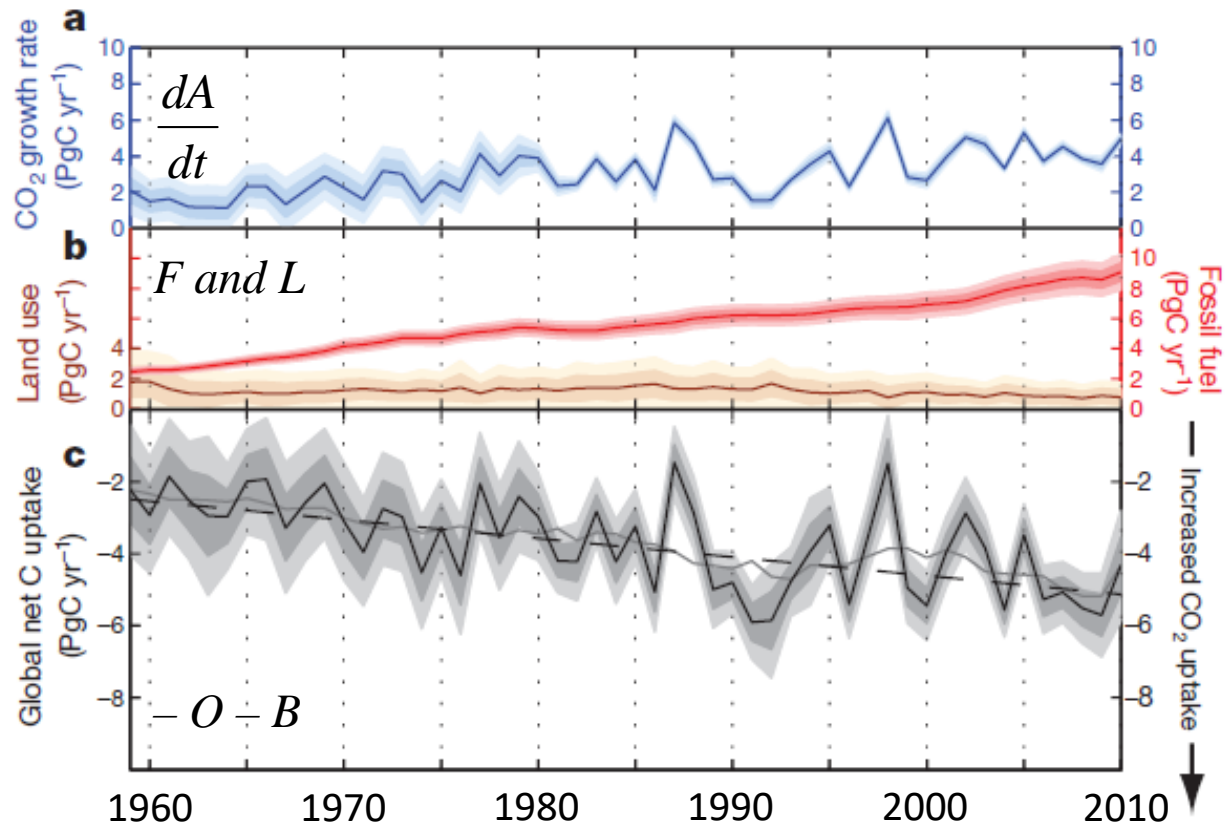
# Atmospheric CO<sub>2</sub> integrates net sources and sinks

$$\frac{dA}{dt} = F + L - O - B$$

Sinks of CO<sub>2</sub> in the land and ocean have been growing consistently with emissions

Airborne fraction: 45%

Total uptake  $5.0 \pm 0.9$  PgC/yr in 2012, doubled since 1960



Ballantyne et al. 2012

# Global Carbon Budget (2005-2014 average)

Fossil Fuels

$33.0 \pm 1.8$  GtCO<sub>2</sub>/yr 91%



Land Use

$3.4 \pm 1.8$  GtCO<sub>2</sub>/yr 9%



Atmosphere

$16.0 \pm 0.4$  GtCO<sub>2</sub>/yr 44%



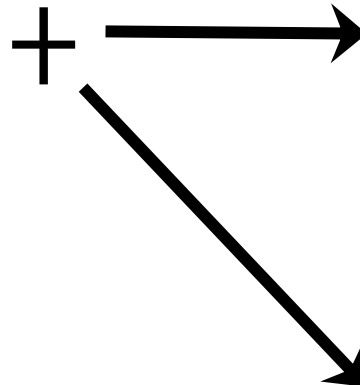
Land Biosphere

$10.9 \pm 2.9$  GtCO<sub>2</sub>/yr 30%



Oceans

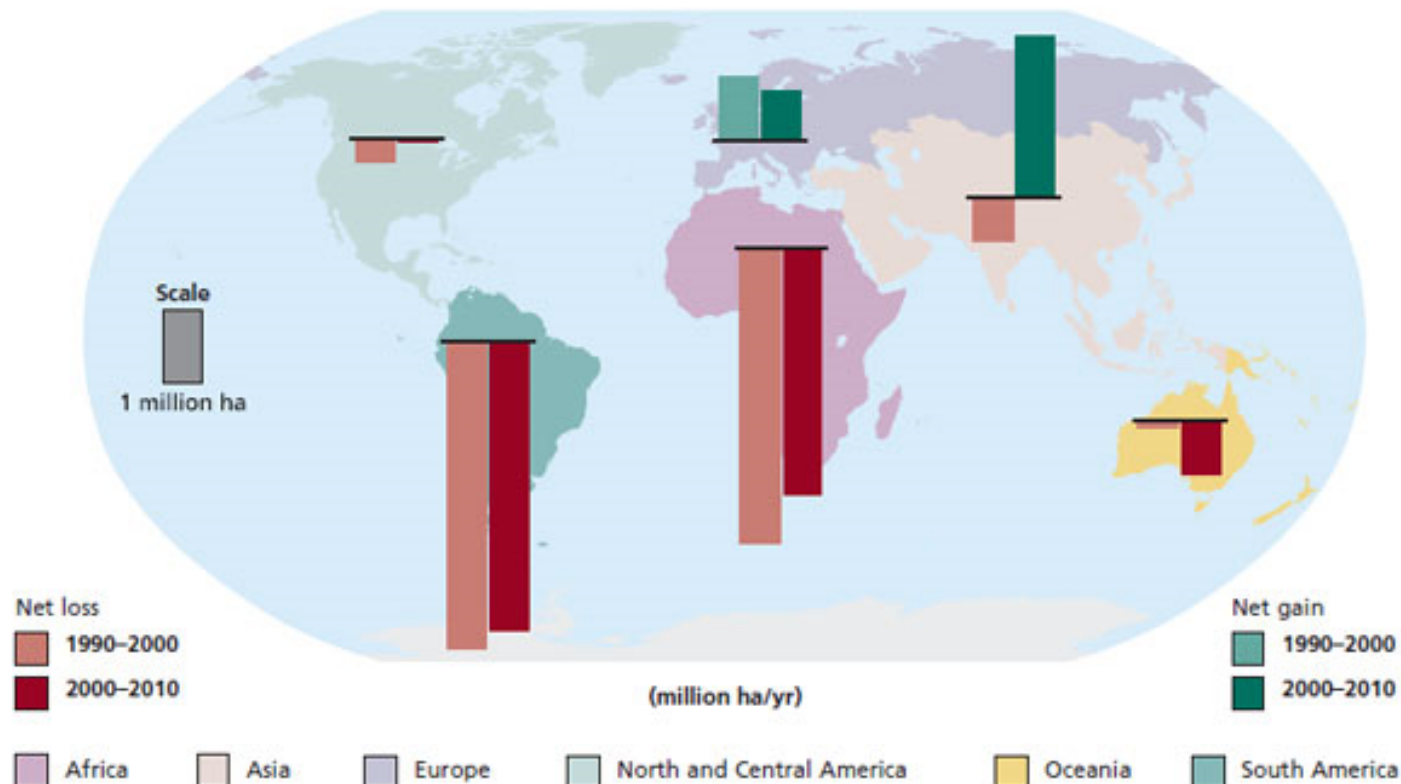
$9.5 \pm 1.8$  GtCO<sub>2</sub>/yr 26%





# Land use is dominated by tropical deforestation emissions, but net uptake in some places

FIGURE 4  
Annual change in forest area by region, 1990–2010

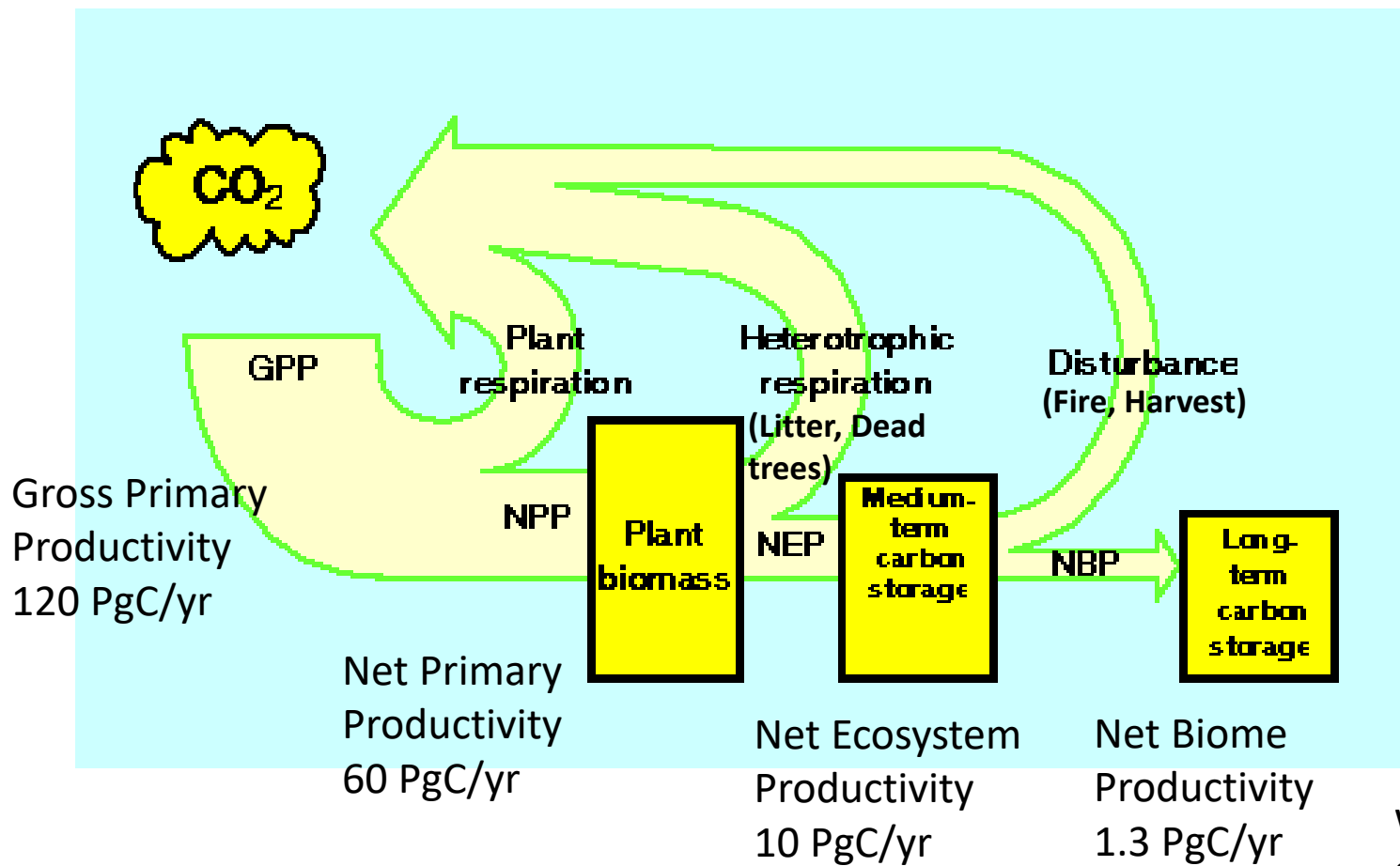




# Observations of terrestrial fluxes and biomass



# CO<sub>2</sub> exchange with the terrestrial biosphere incorporates various processes and timescales

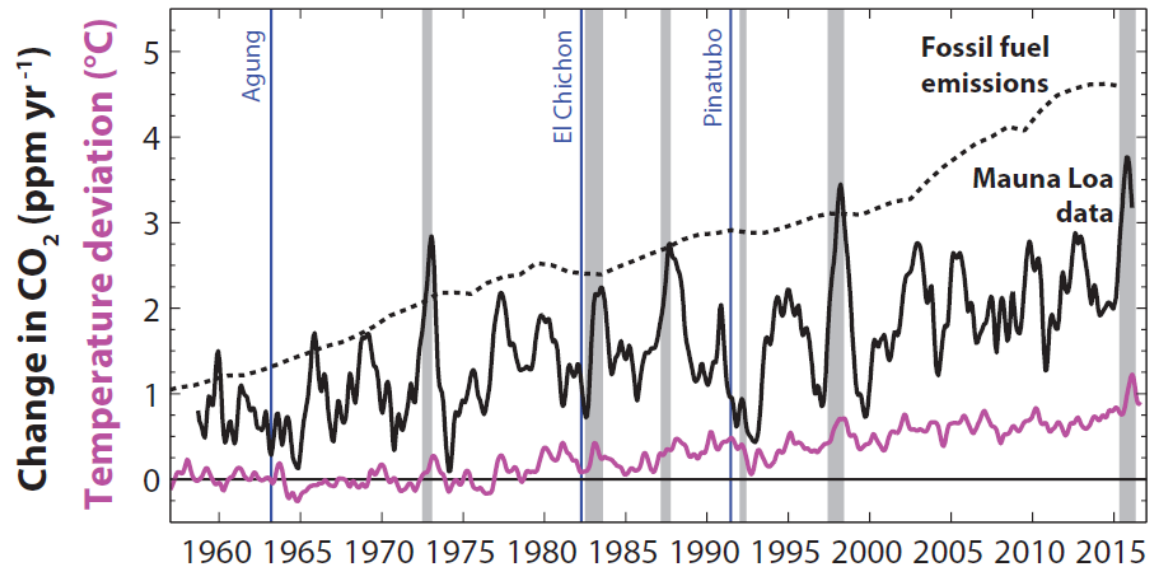
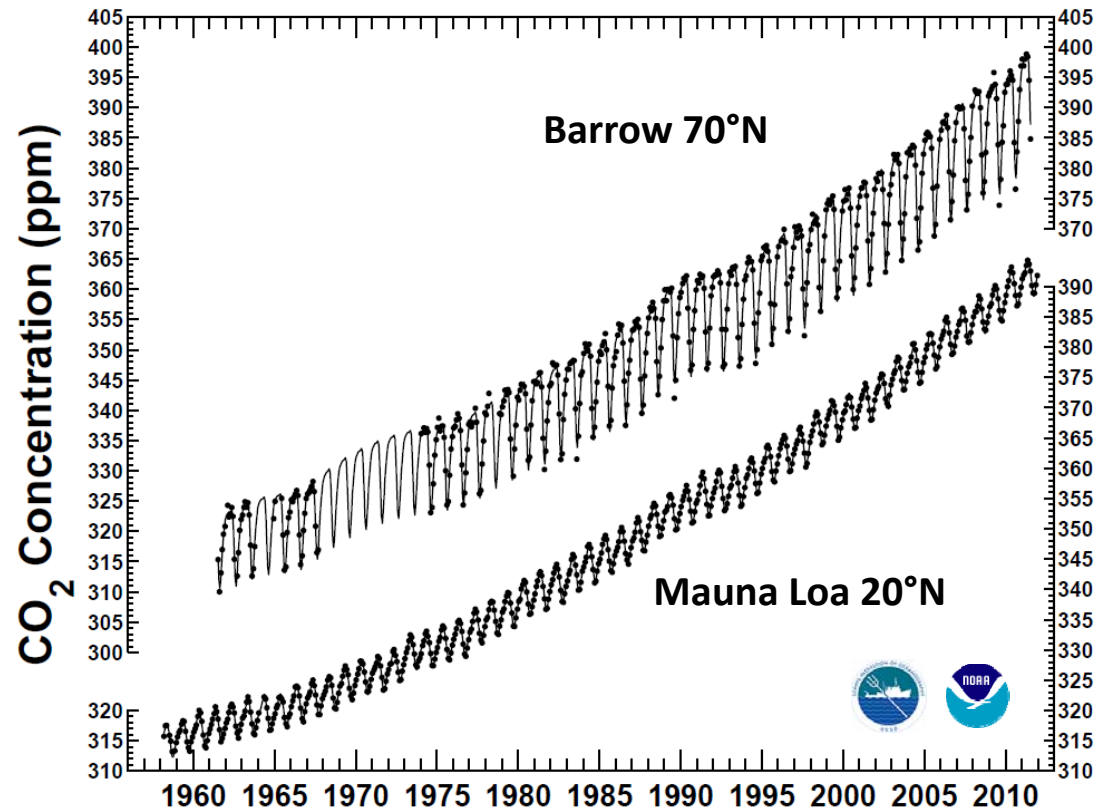


Walker and Steffen 1997;  
IPCC 2013

# Seasonal and interannual variations in biosphere exchange are visible in atmospheric CO<sub>2</sub>

El Niño causes strong release from the biosphere during warm and dry conditions in the tropics

Scripps CO<sub>2</sub> Program; NASA; CDIAC; Graven 2016





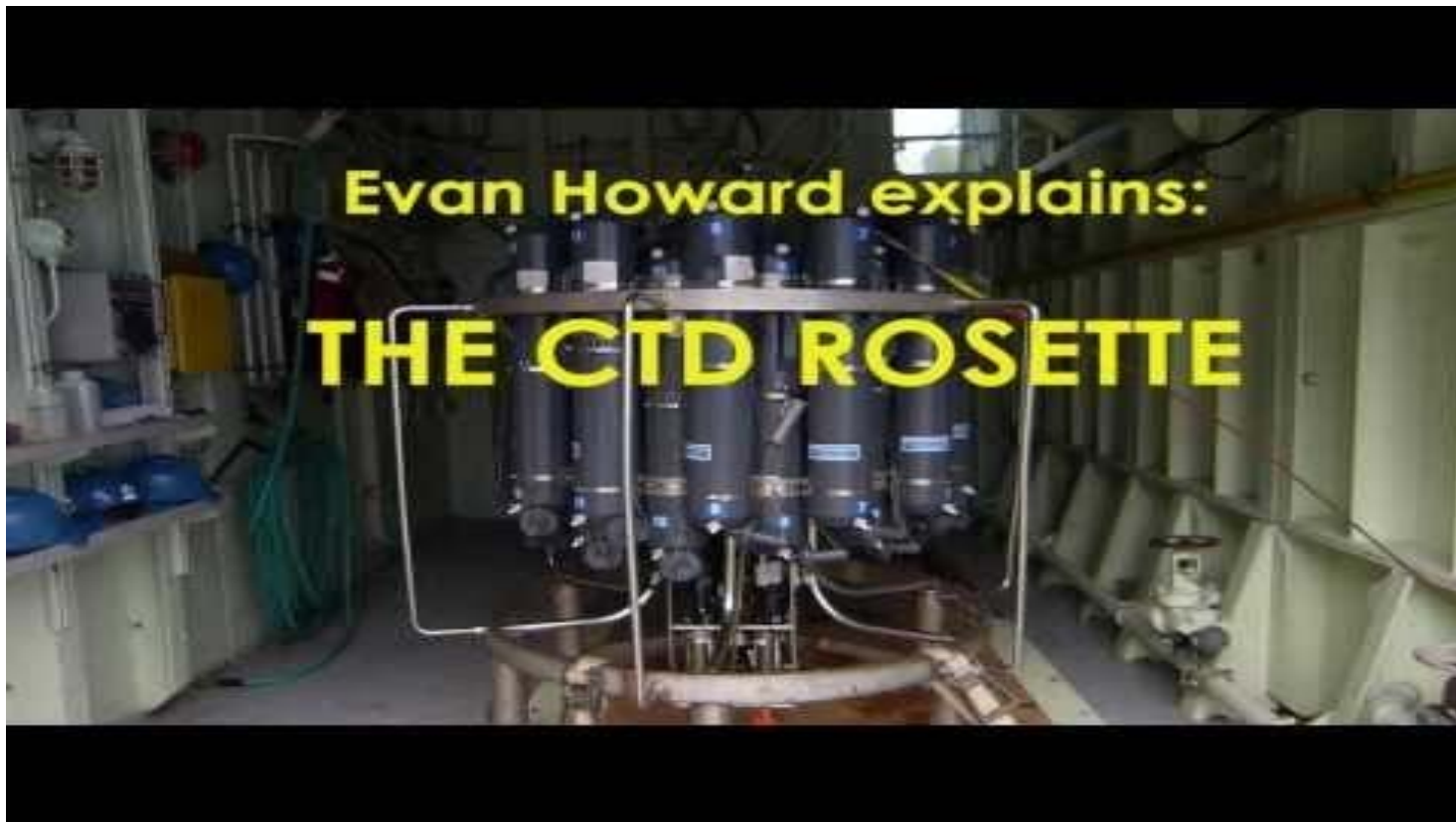
# Oceanic CO<sub>2</sub> and pH measurements use titrations and more advanced technologies



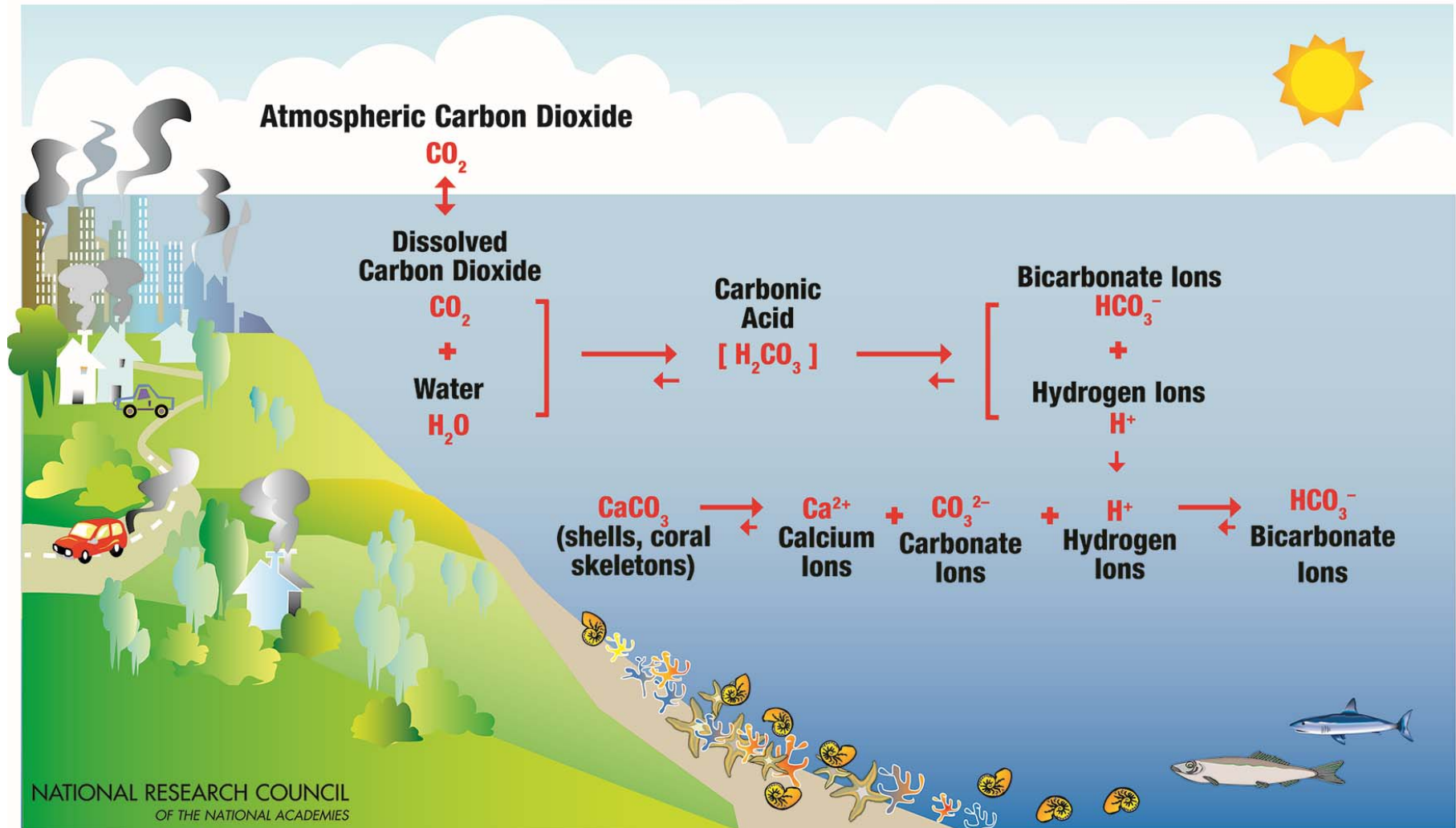
Images courtesy NOAA, NASA

# Measuring conductivity, temperature, and depth (CTD) and sampling ocean water with a Rosette

Video: <https://www.youtube.com/watch?v=7N2UsPDczTw>

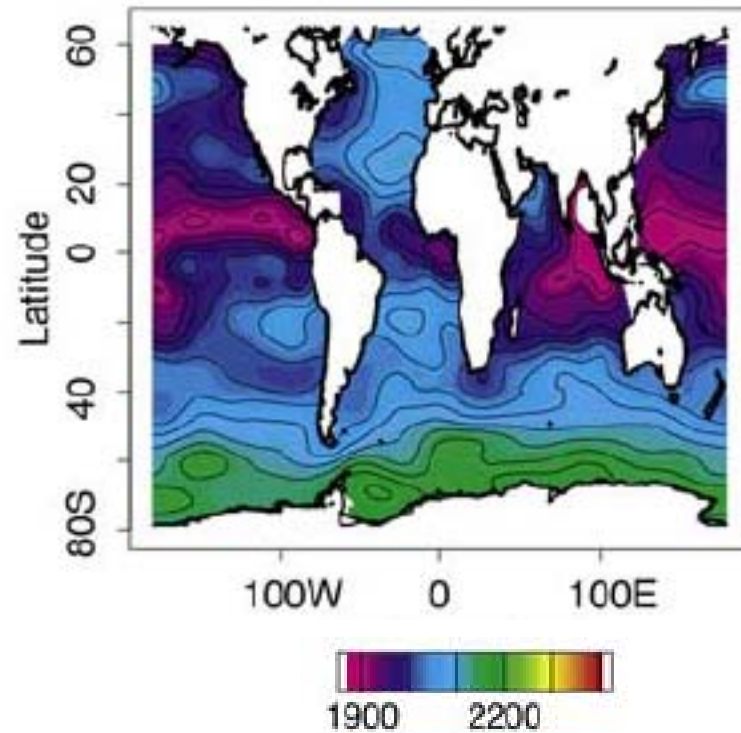
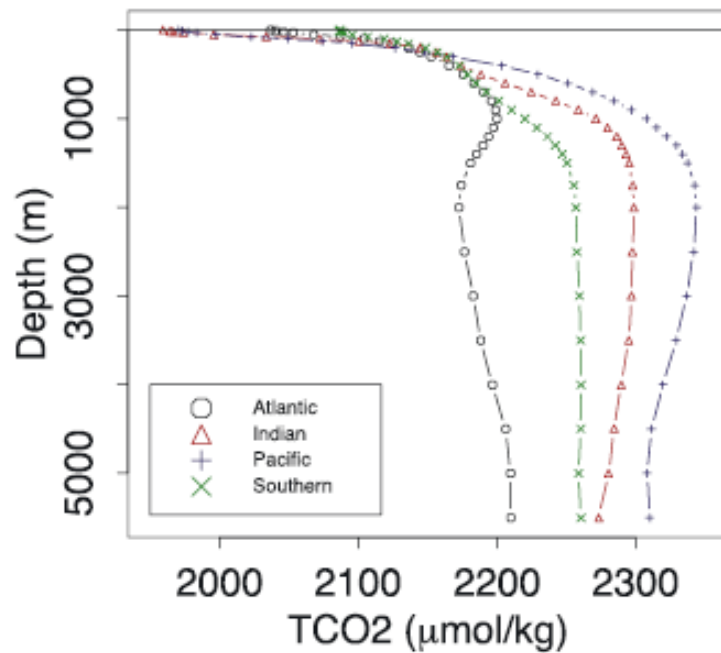


# CO<sub>2</sub> in the ocean takes the form of several carbonate species





# Dissolved Inorganic Carbon is concentrated in the deep ocean because of the solubility and biological pumps

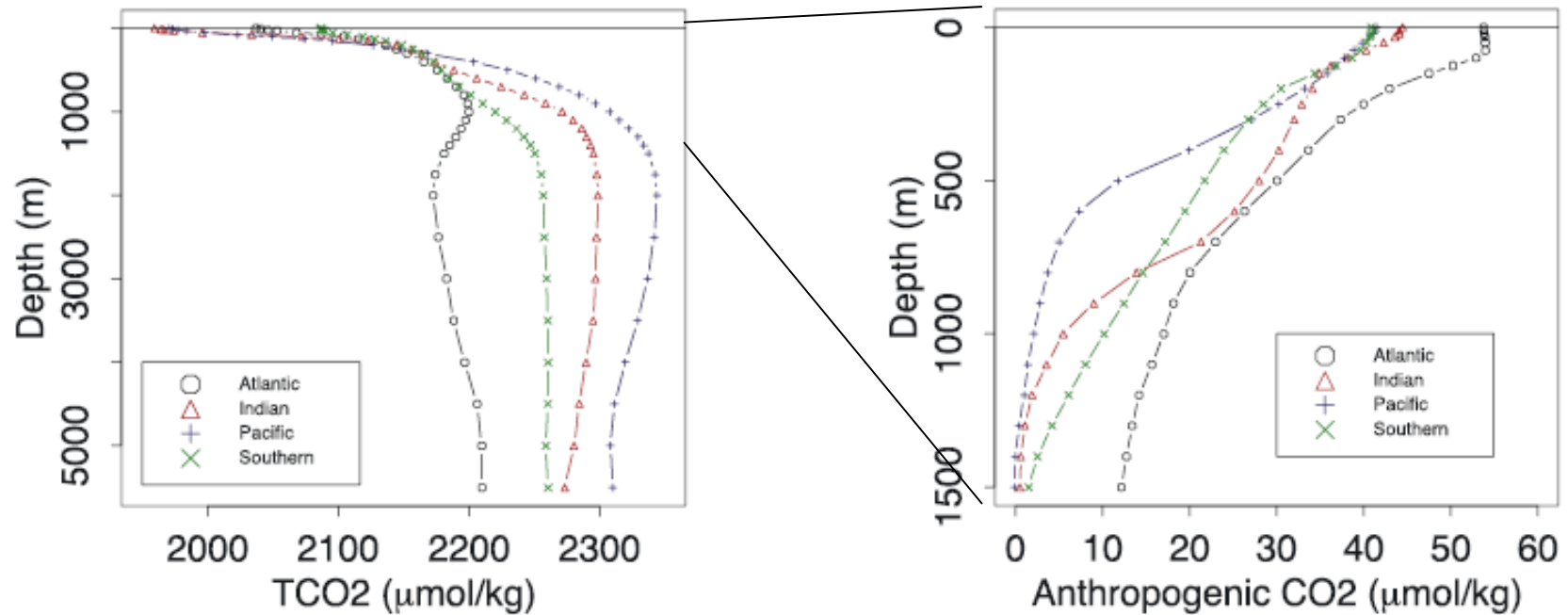


CO<sub>2</sub> solubility in the ocean is mainly a function of TCO<sub>2</sub> and Temperature

Key et al. 2004

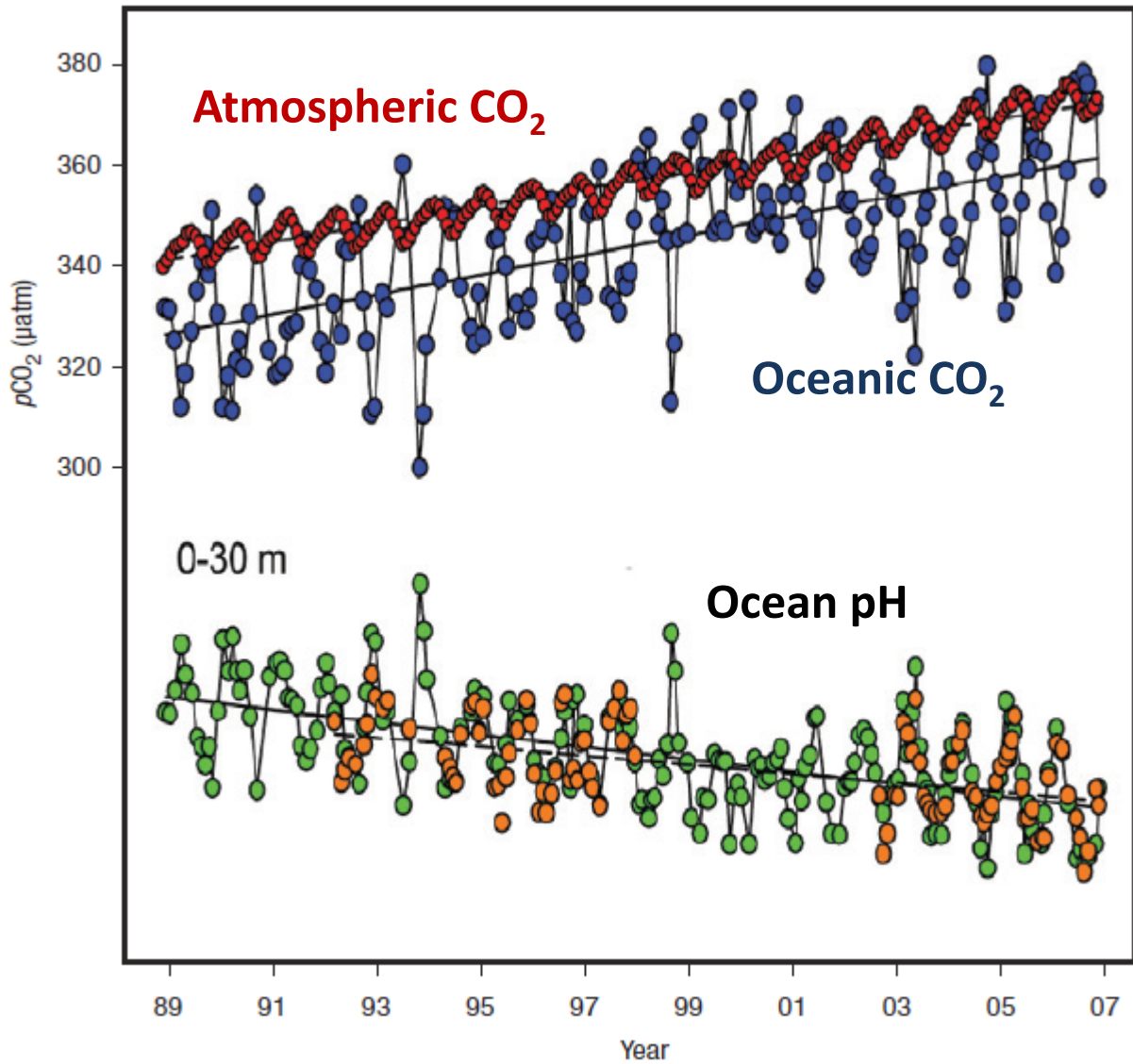


# Anthropogenic CO<sub>2</sub> uptake is driven by dissolution and ocean circulation



Key et al. 2004

Observations show surface ocean CO<sub>2</sub> and H<sup>+</sup> are rising, and pH is decreasing:  $\text{pH} = -\log[\text{H}^+]$



Observations from Station ALOHA, near Hawaii

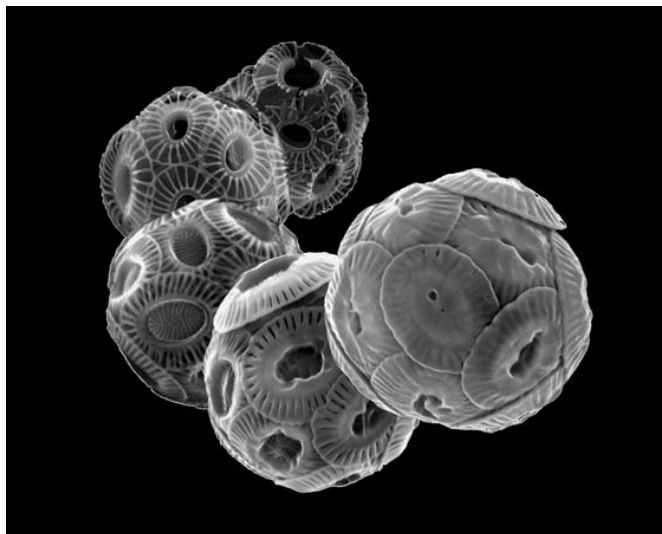
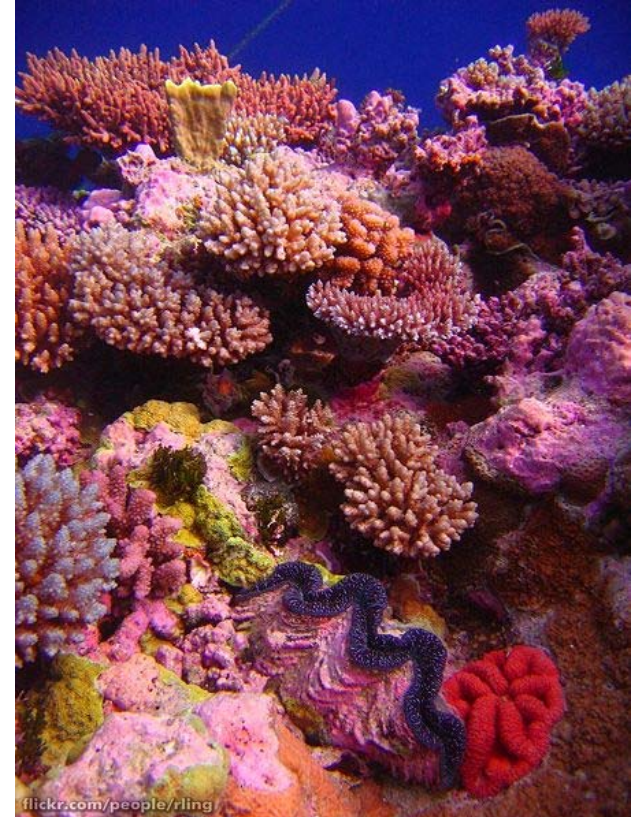
Increasing Acidity

*+25% since the Industrial Revolution*

# Ocean acidification impacts on marine species



Hard corals largely composed of calcium carbonate; Soft corals contain calcium carbonate blended with protein



Calcifying phytoplankton

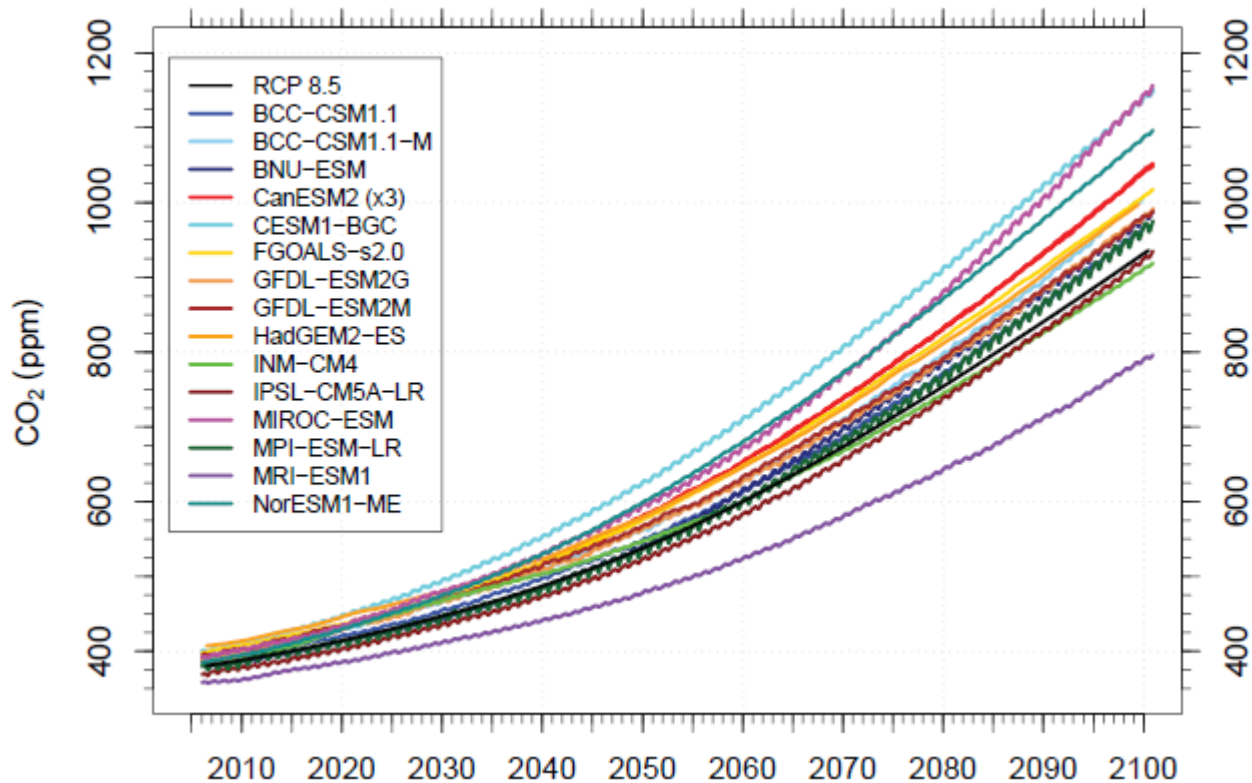
Basal resource in virtually all marine food webs

Vital source of energy for higher trophic levels

From E. O’Gorman

# Carbon cycle processes are one of the main sources of uncertainty in projections of climate change

## ESM RCP 8.5 Atmospheric CO<sub>2</sub>



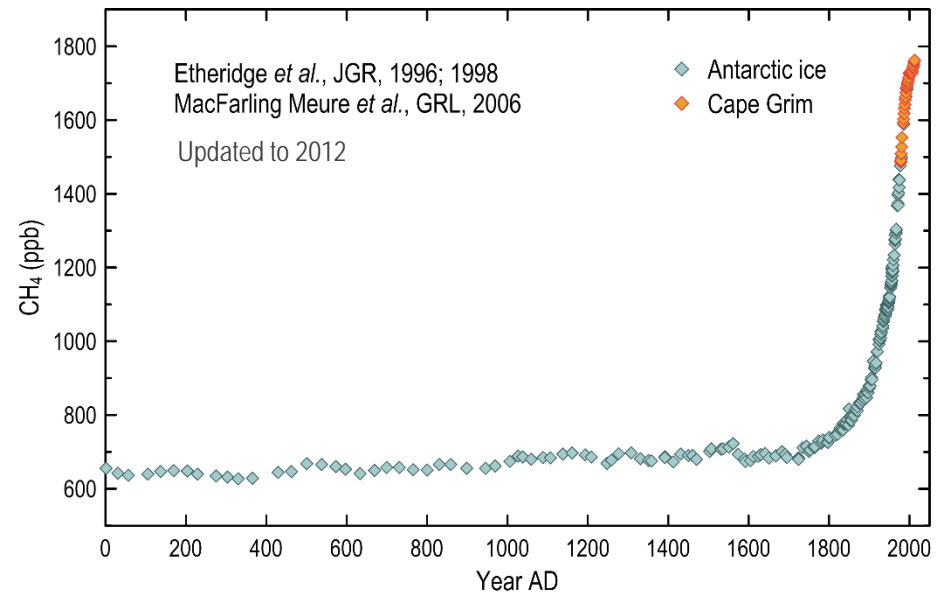
>200 ppm range

Projections of future atmospheric CO<sub>2</sub> under “business-as-usual” RCP8.5 show a wide range between models

Hoffman et al. (2014)

# Atmospheric methane

- After carbon dioxide (CO<sub>2</sub>), methane (CH<sub>4</sub>) is the second most important greenhouse gas contributing to human-induced climate change.
- For a time horizon of 100 years, CH<sub>4</sub> has a Global Warming Potential 28 times larger than CO<sub>2</sub>.
- Methane is responsible for 20% of the global warming produced by all greenhouse gases so far.
- The concentration of CH<sub>4</sub> in the atmosphere is 150% above pre-industrial levels (cf. 1750).
- The atmospheric life time of CH<sub>4</sub> is 9±2 years, making it a good target for climate change mitigation



- Methane also contributes to tropospheric production of ozone, a pollutant that harms human health and ecosystems.
- Methane also leads to production of water vapor in the stratosphere by chemical reactions, enhancing global warming.



# Global methane emissions 2003-2012

## Bottom-up budget

Process models, inventories,  
data driven methods  
**734 TgCH<sub>4</sub>/yr [596-884]**

## Top-down budget

Atmospheric inversions  
**559 TgCH<sub>4</sub>/yr [540-568]**

Mean [min-max range %]

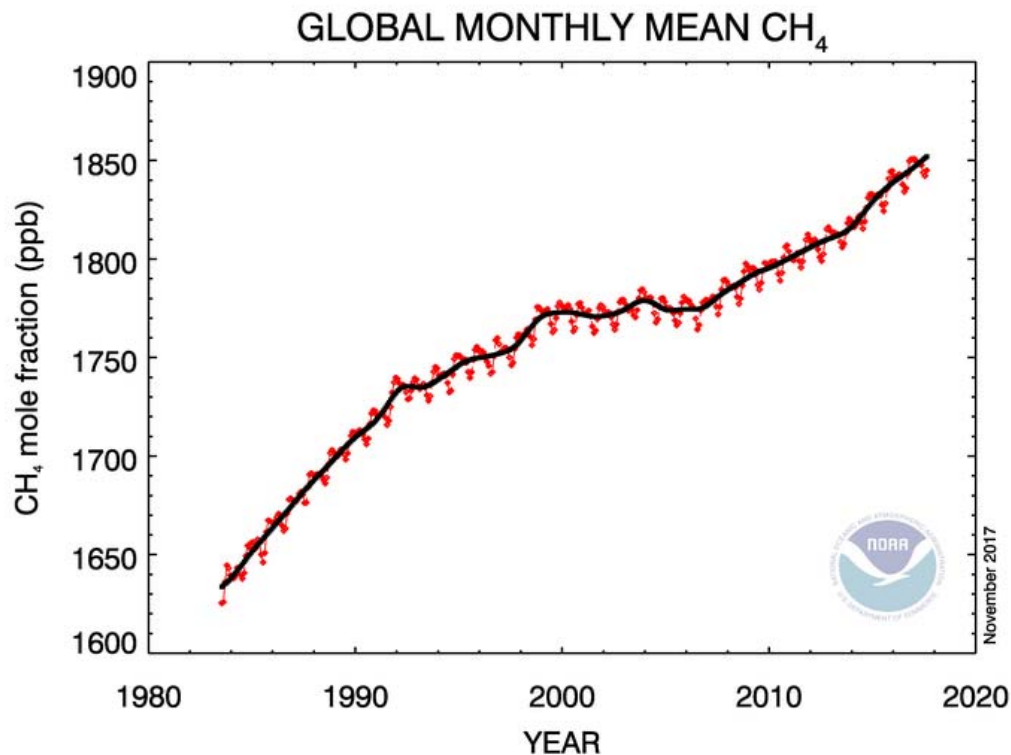
(TgCH<sub>4</sub>/yr)

<b>Natural wetlands</b>	<b>185 [40%]</b>
<b>Agriculture &amp; waste</b>	<b>195 [15%]</b>
Rice	30 [10%]
Enteric ferm & manure	106 [20%]
Landfills & waste	59 [20%]
<b>Fossil fuel use</b>	<b>121 [20%]</b>
Coal	42 [80%]
Gas & oil	79 [10%]
<b>Biomass/biofuel burning</b>	<b>30 [30%]</b>
<b>Other natural emissions</b>	<b>199 [90%]</b>
Fresh waters	122 [100%]
Wild animals	10 [100%]
Wild fires	3 [100%]
Termites	9 [120%]
Geological	40 [50%]
Oceans	3 [100%]
Permafrost	1 [100%]

Mean [uncertainty=  
min-max range %]



# CH<sub>4</sub> concentration stopped increasing for several years in the 2000s, then resumed



## Hypotheses:

- Agricultural emissions (Schaefer et al. 2016)
- Fossil emissions from US fracking (Turner et al. 2016)
- Tropical wetland emissions (Nisbet et al. 2016)
- Variation in atmospheric OH and thus CH<sub>4</sub> removal by OH (Turner et al. 2017; Rigby et al. 2017), perhaps related to CO emissions (Gaubert et al. 2017)



# Major research topics

- Where and how is anthropogenic CO<sub>2</sub> being taken up by the ocean and terrestrial biosphere?
- How will CO<sub>2</sub> and CH<sub>4</sub> exchanges be influenced by future emissions and climate change?
- How are marine organisms and food webs responding to ocean acidification, climate change, deoxygenation?
- How are terrestrial ecosystems responding to changing climate, nutrients, air pollution?
- What are the sources of CH<sub>4</sub> to the atmosphere and what caused the variation in CH<sub>4</sub> growth rate in the 2000s?

# Other reading

Natalie Angier, Too much of a good thing, NYT, Sept. 22, 2014

[http://www.nytimes.com/2014/09/23/science/carbon-dioxide-building-block-of-life-best-in-moderation.html?\\_r=1](http://www.nytimes.com/2014/09/23/science/carbon-dioxide-building-block-of-life-best-in-moderation.html?_r=1)

Heather Graven, The carbon cycle in a changing climate, *Physics Today*.

<http://physicstoday.scitation.org/doi/10.1063/PT.3.3365>

The Global Carbon Project <http://www.globalcarbonproject.org/>

IPCC AR5 Chapter 6: Carbon and Other Biogeochemical Cycles

<http://www.ipcc.ch/report/ar5/wg1/>

David Archer, *The Global Carbon Cycle*, Princeton Primers in Climate, Princeton University Press,

Chapter 1: <http://press.princeton.edu/chapters/s9379.pdf>

# Practice questions

1. Write down and explain the key reactions in the global carbon cycle.
2. What is the difference between gross and net fluxes of carbon? Give some examples of how net fluxes change if they are integrated over different timescales.
3. What is the definition of the airborne fraction and what is its value? In one future scenario, CO<sub>2</sub> emissions will increase to 20 PgC/yr by 2050. What would the atmospheric growth rate of CO<sub>2</sub> be in 2050 if the airborne fraction remained constant? What are some potential changes to ocean and terrestrial biosphere processes that could cause the airborne fraction to increase?